Characterization of the Room-Temperature Structure of SnP₂O₇ by ³¹P Through-Space and Through-Bond NMR **Correlation Spectroscopy**

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The room-temperature structure of SnP_2O_7 has been investigated using various one- and two-dimensional (1- and 2D) magic-angle spinning (MAS) NMR experiments. We show that ³¹P homonuclear through-bond and through-space correlation NMR spectroscopies allow us to discriminate between the 12 possible space group symmetries of the structure. In particular, the ³¹P 2D refocused INADEQUATE experiment allows us to resolve an impressive number of distinct ³¹P resonances, showing that the asymmetric unit contains at least 49 different P₂O₇ groups with two inequivalent P sites. The results suggest that the symmetry of the SnP_2O_7 room-temperature structure is monoclinic with space group $P2_1$ or Pc.

Introduction

Compounds of the $M^{4+}P_2O_7$ family (M = Si, Ge, Sn, Pb, Ti, Zr, Hf, etc.) have recently generated considerable interest because several members of the family show low or even negative coefficients of thermal expansion over certain temperature ranges.^{1–3} At high temperature, these compounds usually have a cubic structure, with Z = 4 and space group symmetry $Pa\bar{3}$, which can be described as a network of corner-sharing MO_6 octahedra and PO₄ tetrahedra linked together to form P_2O_7 units (Figure 1).² In this ideal cubic structure, the P_2O_7 groups are on 3-fold axes with the P-O-P bridging oxygen atoms at inversion centers and the P-O-P bond angles constrained by symmetry to be 180°.² However, the energetically preferred bond angles for these linkages are in the range $130^{\circ}-160^{\circ 2}$ and at low temperature linear P₂O₇ units are partially or completely avoided by distortion of the ideal cubic structure. The bending of these linear bonds results in a pseudocubic structure with a tripled $(3 \times 3 \times 3)$ unit cell (*Z* = 108). This pseudo-cubic structure, usually described as a 3 \times 3 \times 3 superstructure relative to the aristotypic structure ($Pa\bar{3}, Z=4$), has been commonly observed at ambient temperature in such systems by various experimental techniques.^{3–11}

- Laud, K. R.; Hummel, F. A. J. Am. Ceram. Soc. 1971, 54, 296.
 Korthuis, V.; Khosrovani, N.; Sleight, A. W.; Roberts, N.; Dupree,
- R.; Warren, W. W. Chem. Mater. 1995, 7, 412.
- (3) Khosrovani, N.; Korthuis, V.; Sleight, A. W.; Vogt, T. Inorg. Chem. 1996, 35, 485.
- (4) Vollenke, H.; Wittmann, A.; Novotny, H. Monatsh. Chem. 1963, 94, 956.





Figure 1. Ideal cubic structure of SnP₂O₇ shown as cornersharing SnO₆ octahedra and PO₄ tetrahedra.

Although powder and electron diffraction methods give clear evidence of the tripled cell parameters, the exact nature of these room-temperature structures and their space group symmetries is a matter of some debate. Good quality single crystals of these materials can be hard to obtain and the structural complexity makes detailed interpretation of power diffraction patterns difficult. It can be shown that, assuming phase transitions with group-subgroup relationships, the room-temperature structure can adopt 1 of 12 possible

- (8) Sanz, J.; Iglesias, J. E.; Soria, J.; Losilla, E. R.; Aranda, M. A.
- (a) Bull, S., Iglenis, S. E., Born, S., Losin, E. R., Manda, M. A.
 (b) Losilla, E. R.; Cabeza, A.; Bruque, S.; Aranda, M. A. G.; Sanz, J.; Iglesias, J. E.; Alonso, J. A. *J. Solid State Chem.* **2001**, *156*, 213.
- J. BRISHES, J. E., ADDISO, J. A. J. SOUG State Chem. 2001, 156, 213. (10) Withers, R. L.; Tabira, Y.; Evans, J. S. O.; King, I. J.; Sleight, A. W. J. Solid State Chem. 2001, 157, 186. (11) Gover, R. K. B.; Withers, N. D.; Allen, S.; Withers, R. L.; Evans, J. S. O. J. Solid State Chem. 2002, 166, 42.

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[‡] University of Durham.

⁽⁶⁾ Tillmanns, E.; Gebert, W.; Baur, W. H. J. Solid State Chem. 1974, 7, 69.

⁽⁷⁾ Norberg, S. T.; Svensson, G.; Albertsson, J. Acta Crystallogr. 2001, C57, 225.

Table 1. Number of Different Types of P₂O₇⁴⁻ Groups for the Possible Space Groups Adopted at Room Temperature by AP₂O₇ Materials^a

	O site symmetry P–O–P geometry						
space group	$\overline{\frac{\bar{3}}{P_a-O-P_a}}_{linear}$	$ar{1} P_a - O - P_a$ linear	$\begin{array}{c} 3 \\ P_a - O - P_b \\ linear \end{array}$	$\begin{array}{c}1\\P_a-O-P_b\\bent\end{array}$	no. unique P_2O_7 's	no. unique P sites	% linear P ₂ O ₇
$Pa\bar{3}~(1 imes 1 imes 1)$	1(4)				1	1	100
Pa3 P213	1(4)		1(8) 3(12)	4(96) 8(96)	6 11	11 22	11.1 11.1
R3 R3	1(1)	1(3)	1(2) 3(3)	17(102) 35(105)	20 38	38 76	5.6 2.8
Pbca Pca2 ₁		1(4)		13(104) 27(108)	14 27	27 54	3.7 0
$P2_{1}2_{1}2_{1}$ $P2_{1}/c$		2(4)		27(108) 26(104)	27 28	54 54	0 3.7
Pc $P2_1$				54(108) 54(108)	54 54	108 108	0 0
PĪ P1		4(4)		52(104) 108(108)	58 108	108 216	3.7 0

^a The site symmetry adopted by the P-O-P bridging oxygen determines the geometry of each $P_2O_7^{4-}$ unit. For each case the number gives the number of crystallographically unique P_2O_7 groups of a given type and the number in parentheses the total number of such groups in the unit cell. The total cell content is $Zr_{108}(P_2O_7)_{108}$. The total number of unique P sites in the asymmetric unit is given by Σ col1 $- col^2 + 2*col^3 + 2*col^4.$

space groups derived from the symmetry $(Pa\bar{3})$ of the ideal high-temperature phase.² Several members of the family such as $SiP_2O_7^{6,12}$ and $TiP_2O_7^{7,8,13}$ have been shown to adopt a room-temperature cubic structure with space group $Pa\bar{3}$, which retains the space group of the high-temperature phase but triples each of the cubic cell parameters. Other compounds are known to conform to lower symmetry structures. This was first reported in the case of γ -GeP₂O₇, which has a monoclinic structure with symmetry P21/c or lower.9 For ZrP2O7, ³¹P solidstate NMR experiments have shown that the roomtemperature structure is orthorhombic with a Pbca space group.¹⁴ More recently, X-ray, electron, and neutron diffraction studies have shown that roomtemperature structure of SnP₂O₇ is pseudo-cubic with symmetry P2₁3 or lower.¹¹

These various space group symmetries give rise to very subtle differences in the reflections observed by powder X-ray diffraction. However, the number of crystallographically distinct phosphorus sites and the number of different types of P₂O₇ groups in the asymmetric unit dramatically increase by lowering the symmetry of the possible space group (see Table 1).² In addition, the site symmetry adopted by the P-O-P bridging oxygen atom determines the geometry of each $P_2O_7^{4-}$ unit. For example, $Pa\bar{3}$ symmetry is expected to give rise to 11 crystallographically different P sites forming 6 independent P₂O₇ groups, with 4 in general positions, 1 with the bridging oxygen atom on a 3-fold symmetry site, and 1 with the bridging oxygen atom on a $\overline{3}$ symmetry site (constraining the two P atoms of the P_2O_7 group to be crystallographically equivalent and the P-O-P bond angle to 180°). In contrast, the space group Pbca leads to 27 crystallographically distinct P sites corresponding to 13 P₂O₇ groups in general positions and 1 with the bridging oxygen atom on an inversion center, while for Pca21 one would expect 54

different P sites forming 27 P2O7 groups in general positions.

High-resolution solid-state correlation NMR techniques have proved to be very useful in assigning the space groups of several of these MP₂O₇ compounds¹²⁻¹⁴ and other crystalline structure^{15–18} since they allow the number of independent P sites in the asymmetric unit and the phosphorus connectivity scheme to be probed. In the following, we have examined the room-temperature structure of SnP₂O₇ using various one- and twodimensional magic-angle spinning (MAS) NMR experiments. We show that the use of ³¹P homonuclear through-bond and through-space correlation NMR spectroscopy allows discrimination between the 12 possible space group symmetries of the structure.

Experimental Section

The synthesis of a high-purity crystalline sample of SnP₂O₇ has been described in detail elsewhere.11 In short, SnCl4:5H2O and H₃PO₄ were mixed together with a Sn:P ratio of 1:3. The mixture was heated at 2 K/min to 598 K in air at atmospheric pressure and held at this temperature overnight. The resultant powder was reground and heated at 1473 K for 60 h and then air-quenched from the furnace to yield crystalline SnP₂O₇ quantitatively. The high purity of the sample was confirmed by X-ray and neutron powder diffraction.

The solid-state NMR experiments were performed at room temperature on a Bruker DSX 300 spectrometer (7.0 T) with a 4-mm MAS probehead operating at a Larmor frequency of 121.4 MHz for ³¹P. The one-dimensional (1D) ³¹P MAS spectra were recorded at 3, 10, and 15 kHz spinning frequencies using single-pulse ($\pi/8$) acquisition with a recycle delay of 200 s to ensure no saturation.

The two-dimensional (2D) ³¹P through-space single-quantumdouble-quantum (SQ-DQ) MAS NMR correlation spectrum was acquired at 10 kHz spinning frequency. We have used the

⁽¹²⁾ Iuliucci, R. J.; Meier, B. H. J. Am. Chem. Soc. 1998, 120, 9059. (13) Helluy, X.; Marichal, C.; Sebald, A. J. Phys. Chem. 2000, B104, 2836.

⁽¹⁴⁾ King, I. J.; Fayon, F.; Massiot, D.; Harris, R. K.; Evans, J. S. O. Chem. Commun. 2001, 1766.

⁽¹⁵⁾ Nemeth, G.; Pinkerton, A. A.; Stowe, J. A.; Ogle, C. A. Acta Crystallogr. C 1992, 48, 2200.

⁽¹⁶⁾ Davies, J. A.; Dutremez, S. G.; Pinkerton, A. A. Magn. Reson. Chem. 1993, 31, 435.

Sheldrix, G. M.; Heine, A.; Schmidt-Bäse, K.; Pohl, P.; Jones, G.; Paulus, E.; Waring, M. J. *Acta Crystallogr. B* **1995**, *51*, 987.
 Gaudin, E.; Boucher, F.; Evain, M.; Taulelle, F. *Chem. Mater.* 2000, 12, 1715.



Figure 2. (a) POSTC7¹⁹ pulse sequence used for the 2D through-space SQ–DQ correlation experiment, (b) refocused INADEQUATE pulse sequence²⁰ used for the 2D through-bond SQ–DQ correlation experiment, and (c) *z*-filtered spin–echo pulse sequence²² used for the 2D *J*-resolved experiment.

POSTC7 recoupling sequence¹⁹ (Figure 2a) to reintroduce the ³¹P⁻³¹P homonuclear dipolar interaction under MAS conditions and to achieve an efficient broadband excitation and reconversion of double quantum coherences ($\omega_{rt}/2\pi = 69.5$ kHz). Short excitation and reconversion periods of 600 μ s were used. Sixty-four t_1 increments with 16 transients each were collected. The recycle delay was set to 55 s. A presaturation sequence was applied to ensure equivalent conditions for each transient because of the long ³¹P longitudinal relaxation rate ($T_1 \sim 700$ s).

The 2D ³¹P through-bond single-quantum–double-quantum (SQ-DQ) MAS NMR correlation spectrum was acquired while spinning at 10 kHz using the refocused INADEQUATE sequence²⁰ (shown in Figure 2b). The $\pi/2$ pulse length was 2.3 μ s and the delay τ in the excitation and reconversion periods was set to 9 ms. Eighty t_1 increments with 32 transients each were collected using a presaturation sequence and a recycle delay of 55 s.

The 2D homonuclear *J*-resolved spectrum was recorded at a 15 kHz spinning frequency using a *z*-filtered spin–echo sequence (Figure 2c) to obtain pure absorption-mode line shapes.^{21,22} The $\pi/2$ pulse length was 2.3 μ s. Seventy t_1 increments with 32 scans each were collected using a presaturation sequence and a recycle delay of 20 s.

All the two-dimensional pure absorption phase spectra were obtained using a hypercomplex acquisition.²³ To avoid spinning sidebands in the ω_1 dimension, the t_1 time increment was synchronized with the rotor period. The spinning frequency was stabilized to ± 10 Hz for all experiments. ³¹P chemical shifts were referenced relative to 85% H₃PO₄ aqueous solution



Figure 3. (a) ³¹P 1D MAS NMR spectra of SnP_2O_7 recorded at 3 and 10 kHz spinning frequencies. The asterisk marks the center band region. (b) Expansion of the center band region (10 kHz spinning frequency).

using the deshielding convention. Simulations of the 1D MAS and 2D MAS correlation spectra were performed using the dmfit software. 24

The use of a moderate principal magnetic field (7.0 T) allows an efficient averaging of the ³¹P chemical shift anisotropy using MAS spinning frequencies of 10–15 kHz with a large amount of sample. Without the need of ¹H decoupling for this inorganic sample, these spinning frequencies make the rotary resonance condition of the POSTC7 sequence¹⁹ ($\omega_{\rm rf} = 7\omega_{\rm rot}$) easy to achieve. In addition, the line widths of the ³¹P individual resonances in the SnP₂O₇ sample are mainly due to a small distribution of chemical shift as evidenced from the 2D *J*-resolved experiment and, consequently, the 1D MAS and 2D MAS correlation spectra recorded using a higher magnetic field (9.4 T) did not show any significant increase of the spectral resolution.

Results and Discussion

The ³¹P 1D MAS NMR spectra of SnP₂O₇ recorded at 3 and 10 kHz spinning frequencies are depicted in Figure 3. The center band of these spectra (isotropic chemical shift) shows a complex line shape that can be seen to contain more than 17 overlapping ³¹P isotropic resonances, indicating the presence of at least 17 crystallographically different ³¹P sites in the asymmetric unit of SnP₂O₇. In addition to these isotropic resonances, an intense spinning sideband pattern, reflecting the ³¹P chemical shift anisotropy, is observed at low spinning frequency. The average 31 P chemical shift anisotropy, determined from the spinning sideband intensities,²⁵ is characteristic of the distorted PO₄ tetrahedra involved in P₂O₇ groups. In agreement with a previous electron diffraction study, these 1D MAS spectra clearly indicate the exclusion of the possibility of a cubic symmetry with Pa3 space group, for which only 11 independent phosphorus sites are expected. The distinction between the 11 remaining possible space groups containing between 22 (P213) and 216 (P1) individual P sites is more difficult

⁽¹⁹⁾ Hohwy, M.; Jakobsen, H. J.; Eden, M.; Levitt, M. H.; Nielsen, N. C. J. Chem. Phys. **1998**, 108, 2686.

⁽²⁰⁾ Lesage, A.; Bardet, M.; Emsley, L. J. Am. Chem. Soc. 1999, 121, 10987.

⁽²¹⁾ Ernst, R. R.; Bodenhausen, G.; Wokaun, A. *Principles of Nuclear Magnetic Resonance in One and Two Dimensions*; Clarendon Press: Oxford, U.K., 1987.

⁽²²⁾ Brown, S. P.; Pérez-Torralba, M.; Sanz, D.; Claramunt, R. M.; Emsley, L. *Chem. Commun.* **2002**, 1852.

⁽²³⁾ States, D. J.; Haberkorn, R. A.; Ruben, D. J. *J. Magn. Reson.* **1982**, *48*, 286.

⁽²⁴⁾ Massiot, D.; Fayon, F.; Capron, M.; King, I.; Le Calvé, S.; Alonso, B.; Durand, J. O.; Bujoli, B.; Gan, Z.; Hoatson, G. *Magn. Reson. Chem.* **2002**, *40*, 70.

⁽²⁵⁾ Herzfeld, J.; Berger, A. E. J. Chem. Phys. 1980, 73, 6021-6030.



Single Quantum MAS dimension, $\omega_2/2\pi$ (ppm)

Figure 4. ³¹P 2D through-space SQ–DQ MAS correlation spectrum of SnP₂O₇ obtained using the POSTC7¹⁹ sequence and a spinning frequency of 10 kHz. The dashed line indicates the diagonal of the spectrum. The 1D MAS spectrum is shown above the 2D spectrum.

due to the large spectral overlap in this 1D MAS NMR spectrum. However, it should be mentioned that the integrated intensities of the resolved ³¹P resonances located at about -28.9 and -46.8 ppm correspond to approximately 4 and 1% of the total integrated intensity, respectively. Assuming each P site gives rise to approximately the same intensity contribution to the overall envelope of peaks, this suggests the presence of a very large number of P sites in the asymmetric unit and a corresponding low space group symmetry. However, additional experimental information is needed before excluding any of the possible space groups.

As well as the number of individual ³¹P sites, the number of unique P_2O_7 groups in the asymmetric unit can be used for space group assignment. To characterize the number of unique P_2O_7 groups in the SnP_2O_7 structure, we have used the ³¹P 2D through-space SQ-DQ correlation MAS NMR experiment. In this experiment, the coherence transfer between coupled nuclei is driven by the ³¹P-³¹P through-space dipolar interactions leading to the observation of nearest-neighbor spatial proximities. In inorganic pyrophosphates, the intra-P₂O₇ phosphorus-phosphorus interatomic distances are typically 0.3 nm while the inter- P_2O_7 P-P distances are in the range 0.4–0.5 nm. The use of short double-quantum excitation and reconversion periods allows one to differentiate intra-P₂O₇ dipolar coupling and weaker long-range inter-P₂O₇ dipolar interactions and thus to identify the P-O-P connectivity scheme. This has been previously shown in the case of $TiP_2O_{7,13}$ ZrP_2O_7 ,¹⁴ and other crystalline pyrophosphates.^{26,27} The ³¹P 2D through-space correlation MAS spectrum of SnP₂O₇ obtained using the POSTC7¹⁹ sequence is shown in Figure 4. In this SQ–DQ correlation spectrum, the

resonances of two coupled phosphorus atoms occur at a common double-quantum frequency in the ω_1 indirect dimension and are correlated with the single-quantum frequency of each peak in the ω_2 dimension. This leads to the resolution of at least 27 pairs of cross-correlation peaks with various intensities that almost completely overlap in the 1D MAS spectrum. While a large number of these correlation peaks would likely be attributed to the through-space intra- P_2O_7 dipolar connectivities, the presence of additional inter- P_2O_7 longer range dipolar interactions makes the determination of the number of P_2O_7 units in the structure difficult. In this complex case, the observation of the inter-P₂O₇ correlation peaks that reflect the spatial proximity of the different P_2O_7 units damages the spectral resolution of the throughspace correlation spectrum. Nevertheless, at least 38 of the 54 resolved resonances (27 pairs) show significantly different ³¹P isotropic chemical shifts in the singlequantum dimension of the 2D spectrum. This suggests that the asymmetric unit contains at least 38 crystallographically different P sites, ruling out the space groups with <38 independent P sites.

A more convenient approach to determine the throughbond P-O-P connectivities and the number of distinct P_2O_7 groups in the structure would be to use the through-bond scalar J coupling, as routinely done in liquid-state NMR spectroscopy. In rigid solids, the isotropic *J*-coupling is usually much smaller than the dipolar interactions and the chemical shift anisotropy. Nevertheless, ³¹P, ¹³C, or ²⁹Si homonuclear though-bond correlation spectra have already been obtained in solids, 12,13,20,28,29 and recently, we have shown that the refocused-INADEQUATE experiment allows an unambiguous determination of the through-bond P-O-P connectivity scheme in crystalline and disordered phosphates.³⁰ The 2D ³¹P through-bond SQ-DQ correlation MAS spectrum of SnP₂O₇, obtained using the refocused INADEQUATE sequence,²⁰ is depicted in Figure 5. It clearly shows a spectacular gain in resolution relative to the 1D MAS and 2D through-space dipolar correlation spectra. This 2D spectrum exhibits 46 different pairs of resolved cross-correlation peaks that reflect the through-bond intra-P2O7 connectivity. In addition, 3 of the 46 paired correlation peaks display an asymmetric line shape, suggesting the presence of at least two overlapping contributions. To reduce the number of fitting parameters, the experimental SQ-DQ correlation spectrum was sheared along the ω_1 dimension to obtain a symmetric SQ-SQ correlation spectrum with two equivalent dimensions.²¹ As shown in Figure 6, a good fit of the experimental refocused INADEQUATE spectrum was obtained with 98 distinct ³¹P resonances of nearly Lorentzian shape (30% Gaussian admixture) corresponding to 49 P₂O₇ units with two inequivalent P sites. Consequently, all the space groups with <98 independent P sites can be excluded. In the 2D spectrum, the line widths of the individual ³¹P resonances range between 55 and 90 Hz, suggesting that the broader lines of higher intensity could result from additional peak overlaps. It should be noted that the

⁽²⁶⁾ Feike, M.; Graf, R.; Schnell, I.; Jager, C.; Spiess, H. W. J. Am. Chem. Soc. 1996, 118, 9631.
(27) Fayon, F.; Bessada, C.; Coutures, J. P.; Massiot, D. Inorg.

⁽²⁷⁾ Fayon, F.; Bessada, C.; Coutures, J. P.; Massiot, D. Inorg. Chem. **1999**, *38*, 5212.

⁽²⁸⁾ Fyfe, C. A.; Feng, Y.; Gies, H.; Grondey, H.; Kokotailo, G. T. J. Am. Chem. Soc. **1990**, *112*, 3264.

 ⁽²⁹⁾ Wu, G.; Wasylishen, R. E. Organometallics 1992, 11, 3242.
 (30) Fayon, F.; Le Saout, G.; Emsley, L.; Massiot, D. Chem. Commun. 2002, 1702.



Figure 5. ³¹P 2D through-bond SQ–DQ MAS correlation spectrum of SnP_2O_7 obtained at 10 kHz spinning frequency using the refocused INADEQUATE²⁰ experiment. The dashed line indicates the diagonal of the spectrum. The 1D MAS spectrum is shown above the 2D spectrum.

isotropic ${}^{2}J({}^{31}P-{}^{31}P)$ coupling (that drives the coherence transfer in the refocused INADEQUATE experiment) is significantly smaller than these line widths and is not resolved as a doublet in the 2D spectrum. Evidence for this coupling can be clearly shown by the 2D *J*-resolved experiment,^{21,22} which has been previously used to measure weak ${}^{2}J(P-P)$ couplings in metal phosphine complexes. ^{29,31} The 2D J-resolved MAS spectrum of SnP₂O₇ is shown in Figure 7. In this 2D map, the conventional MAS spectrum is retained along the ω_2 dimension while the homonuclear J couplings between two ³¹P nuclei appear along the ω_1 dimension. In the ω_1 dimension, the line width is about 7 Hz, allowing the clear resolution of the characteristic doublet due to the weak ${}^{2}J(P-P)$ coupling and showing that the line width of the individual resonance in MAS spectra is mainly due to a small distribution of chemical shift. From the observed splittings, the intra- $P_2O_7 {}^2 J(P-$ P) coupling constants are estimated to range between 16 and 19.5 Hz, similar to the values measured in SiP₂O₇¹² and TiP₂O₇¹³ using the TOBSY³² experiment.

As mentioned above, the refocused INADEQUATE spectrum of SnP₂O₇ reveals the presence of at least 49 P₂O₇ units with two inequivalent P sites in the asymmetric unit, allowing all the space groups with <98 independent P sites to be ruled out. Only four space groups of lower symmetry remain possible: $P2_1$, Pc, $P\overline{1}$, for which 108 crystallographically distinct P sites in general positions are expected, and P1, which gives rise to 216 different phosphorus sites in general positions. For $P2_1$, Pc, and P1 space groups, no linear P₂O₇ units are required by symmetry, while for $P\overline{1}$, four linear P₂O₇ units with two equivalent P sites (related by a center of inversion) are expected.² Such linear P₂O₇ groups with two magnetically equivalent ³¹P nuclei are not



Figure 6. (a) Sheared²¹ and symmetrized experimental ³¹P 2D refocused INADEQUATE spectrum of SnP_2O_7 and (b) its simulation with 98 distinct ³¹P resonances corresponding to at least 49 P_2O_7 units with two inequivalent P sites.

observable in the refocused INADEQUATE spectrum, but they should give rise to diagonal autocorrelation peaks in a through-space dipolar correlation spectrum, as previously observed in the case of ZrP_2O_7 .¹⁴ Such autocorrelation peaks were not clearly evidenced in the 2D through-space dipolar correlation spectrum of SnP₂O₇ (Figure 4), making the possibility of space group $P\overline{1}$ unlikely. Before one of the remaining possibilities is adopted, the relative intensities of the 98 distinct resonances should be determined. The quantitative 1D MAS spectrum (10 kHz spinning frequency) was thus fitted with the 49 pairs of coupled resonances determined from the 2D refocused INADEQUATE spectrum. The position and line width of each resonance were fixed and only the intensity was allowed to vary. A reasonable fit of the experimental 1D MAS spectrum (not shown) was obtained with 90 peaks of relative intensities in the range 0.85-1.15, 6 peaks of relative intensities in the range 1.9-2.1, and 2 peaks of relative intensities 3.0. These higher intensity peaks correspond to the strong ones in the 2D refocused INADEQUATE spectrum and

⁽³¹⁾ Wu, G.; Wasylishen, R. E. Inorg. Chem. 1992, 31, 145.

⁽³²⁾ Baldus, M.; Meier, B. H. J. Magn. Reson. 1996, A121, 65.



Figure 7. ³¹P 2D *J*-resolved MAS spectrum of SnP_2O_7 recorded at 15 kHz spinning frequency using a *z*-filtered spin– echo²² sequence. The 1D MAS spectrum is shown above the 2D spectrum.

are attributed to the presence of several overlapping contributions having very similar frequencies in both the single-quantum and the double-quantum dimensions. Within experimental error, this is consistent with the space groups $P2_1$ and Pc, for which 108 crystallographically inequivalent P sites are expected. These relative intensities would also be in agreement with space group P1, but this would correspond to a significantly higher degree of spectral overlap in the refocused INADEQUATE spectrum. These results thus suggest that the room-temperature structure of SnP_2O_7 is monoclinic with $P2_1$ or Pc symmetry. The resolution of the experimental X-ray diffraction powder pattern of SnP_2O_7 does not allow one to distinguish between these two possibilities since the three cell parameters are almost equal and Pawley fits³³ of the X-ray diffraction data using $P2_1$ or Pc space groups give very similar reliability factors (Rwp) of 3.19% and 3.23%, respectively.

Conclusion

In this study, we have examined the room-temperature structure of SnP_2O_7 using various solid-state magic-angle spinning NMR experiments. We show that the use of ³¹P homonuclear through-bond and throughspace correlation NMR spectroscopy allows discrimination between the 12 possible space group symmetries of the structure. In particular, the ³¹P 2D refocused INADEQUATE experiment provides evidence of an impressive number of distinct ³¹P resonances and shows that the asymmetric unit contains at least 49 different P_2O_7 groups corresponding to 98 independent P sites, some of which contain accidental degeneracies. These results suggest that the true symmetry of the SnP_2O_7 room-temperature structure is monoclinic with space group $P2_1$ or Pc.

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Supporting Information Available: Table of the ³¹P isotropic chemical shifts, line widths and relative cross-correlation peak intensities of the 98 distinct resonances obtained from the simulation of the 2D refocused INAD-EQUATE spectrum of SnP₂O₇ (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽³³⁾ Pawley, G. S. J. Appl. Crystallogr. 1981, 14, 357.